

COMPARISON OF THREE METHODS TO ANALYZE NON- AROMATIC ORGANIC ACIDS IN HONEY

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IMPORTANCE OF THE DETERMINATION OF ORGANIC ACIDS IN HONEY

(Mato et al. 2003. J. Food Prot. 66)

Organic acids → Minor constituents of honey

Important contributions to honey properties

Antibacterial activity → **Free and total acidity**

(Bogdanov. 1997. Lebensm. Wiss. Technol. 30)

Antioxidant activity → **Organic acids**

(Gheldof et al. 2002. J. Agric. Food Chem. 50)

Indicators of fermentation → **Acetic acid**

(Gonnet. 1982. Opida INRA 2nd ed.)

Treatment against *Varroa* infestation →

(Gregorc and Planinc. 2001. Apidologie 32)

Formic acid
Lactic acid
Oxalic acid

Factors for the characterization of botanical and geographical origins

(Anklam. 1998. Food Chem. 63)

PURPOSE:

To compare **enzymatic**, **HPLC** and **capillary zone electrophoresis (CZE)** procedures to analyze some non-aromatic organic acids in floral honeys.

Precision

Recovery

Specificity

Sensitivity

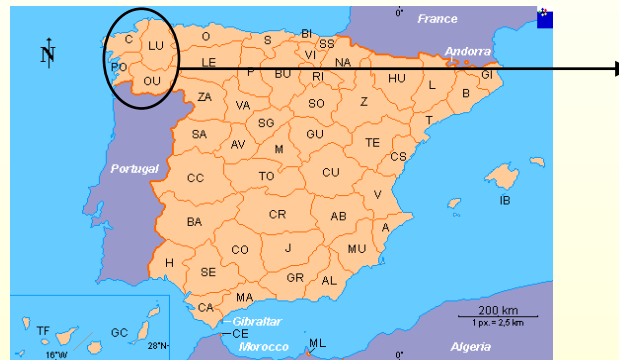
Simplicity

Speed

Cost

GEOGRAPHICAL ORIGIN OF THE SAMPLES

50 Samples of honey



NW SPAIN

GALICIA

BOTANICAL ORIGIN OF THE SAMPLES

Melissopalynology

Honey sediment:

glycerined
solution

methyl-green

hydroalcoholic

Terradillos et al. 1994. Bee Science 3

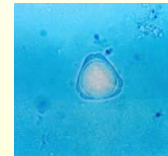
Identification, counting and presentation
of frequencies

Louveaux et al. 1978. Bee World 59

Von der Ohe et al. 2004. Apidologie 35

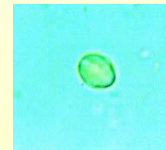
frequency classes

25 multifloral honeys

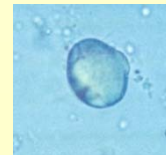
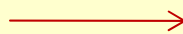


21 eucalyptus honeys (*Eucalyptus* sp.)

3 chestnut honeys (*Castanea sativa*)



1 clover honey (*Trifolium* sp.)





ENZYMATIC METHODS

ENZYMATIC METHODS

Measurements at 340 nm

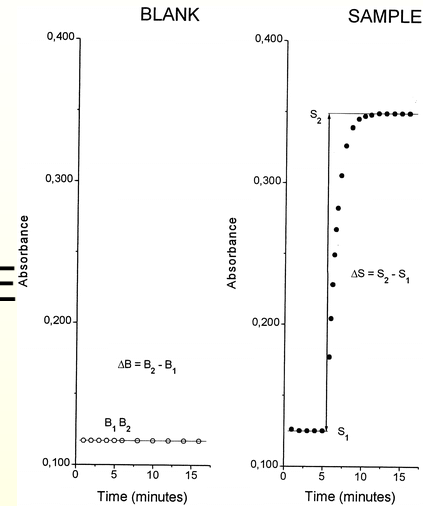
Total D-Gluconic acid

(Mato et al. 1997. J. Agric. Food Chem. 45)

Gluconate kinase

6-phosphogluconate dehydrogenase

NICOTINAMIDE ADENINE DINUCLEOTIDE PHOSPHATE



Citric acid

(Mato et al. 1998. J. Agric. Food Chem. 46)

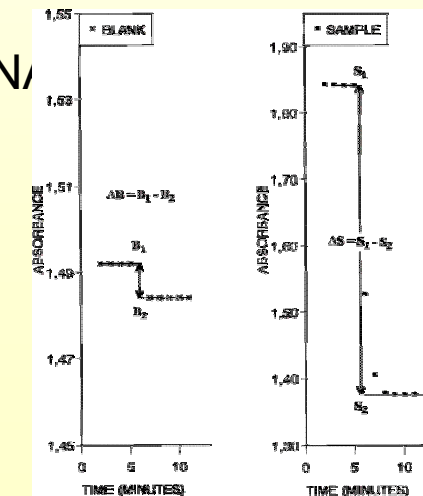
Citrate liase

L-malate dehydrogenase

L-lactate dehydrogenase

REDUCED NICOTINAMIDE ADENINE DINUCLEOTIDE (NAD)

Previous clarification with polyvinylpolypyrrolidone (PVPP)



L-Malic acid

(Mato et al. 1998. Food Chem. 62)

Glutamate-oxaloacetate transaminase

L-malate dehydrogenase

NICOTINAMIDE ADENINE DINUCLEOTIDE (NAD)

ENZYMATIC METHODS

Total D-Gluconic acid

Precision

	HONEY SAMPLE		
	A	B	C
	3.90	7.28	11.72
	3.92	7.28	11.71
	3.92	7.25	11.70
	3.92	7.25	11.71
	3.90	7.28	11.72
	3.92	7.26	11.72
	3.92	7.29	11.71
	3.90	7.28	11.70
	3.90	7.28	11.72
	3.93	7.25	11.70
mean	3.91	7.27	11.71
SD ^a	0.0116	0.0156	0.00876
% CV ^b	0.30	0.22	0.07

^a Standard deviation. ^b Coefficient of variation.

Recovery

present	added (g/kg)	found (g/kg)	recovery (%)
	2.00	3.96	100.0
	2.00	3.97	100.5
	2.00	3.96	100.0
	4.00	6.00	100.3
	4.00	5.99	100.0
	4.00	5.94	99.8
1.96			
	8.00	9.94	99.8
	8.00	9.91	99.4
	8.00	9.90	99.3
	10.00	11.90	99.4
	10.00	11.88	99.2
	10.00	11.94	99.8
mean			99.8
SD ^a			0.403
% CV ^b			0.40

^a Standard deviation. ^b Coefficient of variation.

RESULTS (g/kg)

Mean: 7.37

Standard deviation: 2.92

Spread of values: from 2.38 to 13.53

ENZYMATIC METHODS

Citric acid

Precision: Study of precision of the determination of citric acid (mg/kg), using the direct enzymatic method and the enzymatic method with previous clarification with PVPP

Honey samples

	A		B		C	
	direct	clarified	direct	clarified	direct	clarified
<i>n</i>	10	10	10	10	10	10
mean	44.9	44.2	424.0	428.4	817.2	827.0
SD ^a	1.19	0.707	6.497	2.198	8.323	2.186
% CV ^b	2.66	1.60	1.53	0.51	1.02	0.26

^a Standard deviation. ^b Coefficient of variation.

Recovery: Study of the recovery of the determination of citric acid (mg/kg), using the direct enzymatic method and the enzymatic method with previous clarification with PVPP

Sample A added (mg/kg)	Recovery (%)	
	direct	clarified
25	104.0	99.6
175	100.7	100.4
375	99.8	100.9
775	97.3	100.9
mean	100.5	100.5
SD ^a	2.769	0.614
% CV ^b	2.76	0.61

^a Standard deviation. ^b Coefficient of variation.

RESULTS (mg/kg)

Mean: 116.3

Standard deviation: 116.0

Spread of values: from 20.7 to 451.2

ENZYMATIC METHODS

Malic acid

Precision: Study of precision of the determination of malic acid (mg/kg)

	Honey samples			
	A	B	C	D
<i>n</i>	10	10	10	10
mean	94	240	463	596
SD ^a	3.31	5.39	5.17	4.60
% CV^b	3.5	2.2	1.1	0.8

^aStandard deviation. ^bCoefficient of variation.

RESULTS (mg/kg)

Mean: 91.0

Standard deviation: 132.0

Spread of values: from 8.0 to 578.0

Recovery: Malic acid (mg/kg)

present (mg/kg)	added (mg/kg)	found (mg/kg)	recovery (%)
	50	99	104
	50	95	96
	50	99	104
	250	288	96
	250	300	101
	250	305	103
47	450	480	96
	450	511	103
	450	485	97
	650	681	98
	650	669	96
	650	713	102
<i>n</i>			12
Mean			100
SD ^a			3.45
% CV^b			3.5

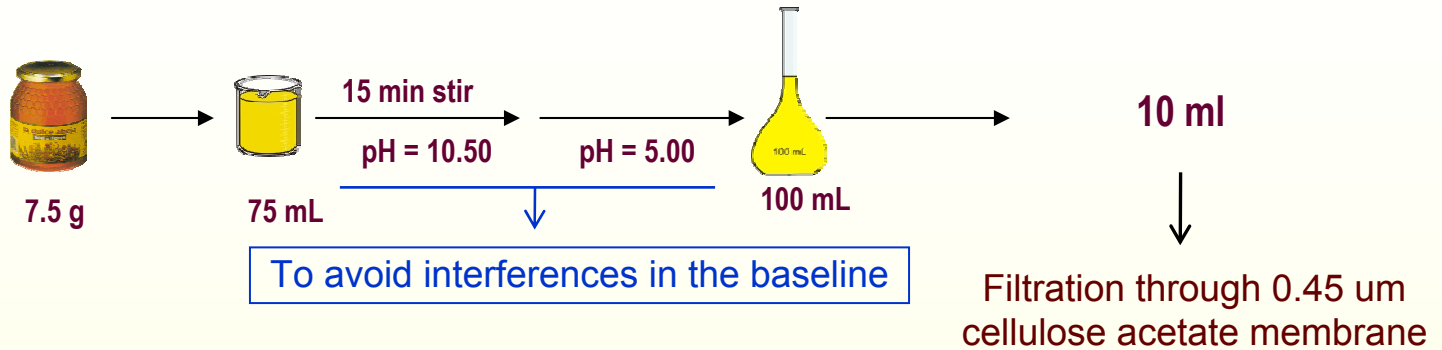
^a Standard deviation. ^b Coefficient of variation.



HPLC PROCEDURE

(Suárez-Luque et al. 2002. J. Chrom A 955)

HPLC PROCEDURE



SOLID-PHASE EXTRACTION (SPE) ANION-EXCHANGE CARTRIDGE

Activation: 10 ml NaOH 0.1 M
percolation rate: 3 ml/min

Sample: **10 ml**
flow-rate: 0.5 ml/min

Cartridge washing: 10 ml water (3 ml/min)

Elution of organic acids: 4 ml H₂SO₄ 0.1 M (0.5 ml/min)

SPE

20 ul

Triplicate



HPLC PROCEDURE

CHROMATOGRAPHIC CONDITIONS

Column: Spherisorb ODS-2 S5

Temperature: 25 °C

Mobile phase: Metaphosphoric acid (pH 2.20)

Flow-rate: 0.7 ml/min

Detection: 215 nm

Time of analysis: **15 minutes**

ORGANIC ACIDS DETERMINED

Malic

Maleic

Citric

Succinic

Fumaric



HPLC PROCEDURE

Retention times

Organic acid	Retention time (min)±SD
Malic	5.81 ± 0.01
Maleic	9.07 ± 0.10
Citric	10.87 ± 0.09
Succinic	11.81 ± 0.10
Fumaric	13.61 ± 0.10

Parameters and correlation coefficients (r) of calibration plots ($y = ax + b$)

$y =$ peak height

$x =$ amount of acid (mg/kg)

Organic acid	a	b	r
Malic	19.79	45.79	1.0000
Maleic	1272	41.66	1.0000
Citric	19.65	0.5058	0.9999
Succinic	6.984	46.70	0.9994
Fumaric	1401	38.05	0.9998

Detection and quantification limits

Organic acid	Detection limit (mg/kg)	Quantification limit (mg/kg)
Malic	1.55	2.93
Maleic	0.059	0.075
Citric	1.44	2.72
Succinic	7.57	10.93
Fumaric	0.0064	0.025

HPLC PROCEDURE

Precision

Repeatability

Acid	Repeatability (n=5)					
	Sample 1		Sample 2		Sample 3	
	Mean±SD (mg/kg)	RSD (%)	Mean±SD (mg/kg)	RSD (%)	Mean±SD (mg/kg)	RSD (%)
Malic	35,5±1,9	2,29	109,4±2,9	2,67	274,1±8,8	3,20
Maleic	0.213±0.011	0.93	0.257±0.008	3.11	0.143±0.003	1.93
Citric	70,9±1,6	0,90	120,8±0,8	0,70	390,3±10,5	2,68
Succinic	23,44±0,04	0,27	31,2±0,22	0,71	152,9±3,1	2,02
Fumaric	0.130±0.008	1.21	1.011±0.007	0.68	6.88±0.24	2.94

Reproducibility

less than

Acid	Reproducibility (n=3)					
	Sample 1		Sample 2		Sample 3	
	Mean±SD (mg/kg)	RSD (%)	Mean±SD (mg/kg)	RSD (%)	Mean±SD (mg/kg)	RSD (%)
Malic	34,6±1,0	2,95	113,1±3,4	3,03	268,0±10,4	3,90
Maleic	0.203±0.008	3.72	0.252±0.011	4.40	0.147±0.004	2.57
Citric	69,5±3,2	4,68	122,3±3,2	2,59	378,4±12,0	3,18
Succinic	23,10±1,12	4,86	30,69±0,92	2,99	149,4±3,4	3,07
Fumaric	0.128±0.005	4.15	0.987±0.045	4.59	7.08±0.09	3.23

HPLC PROCEDURE

Standard solutions recoveries after solid-phase extraction procedure

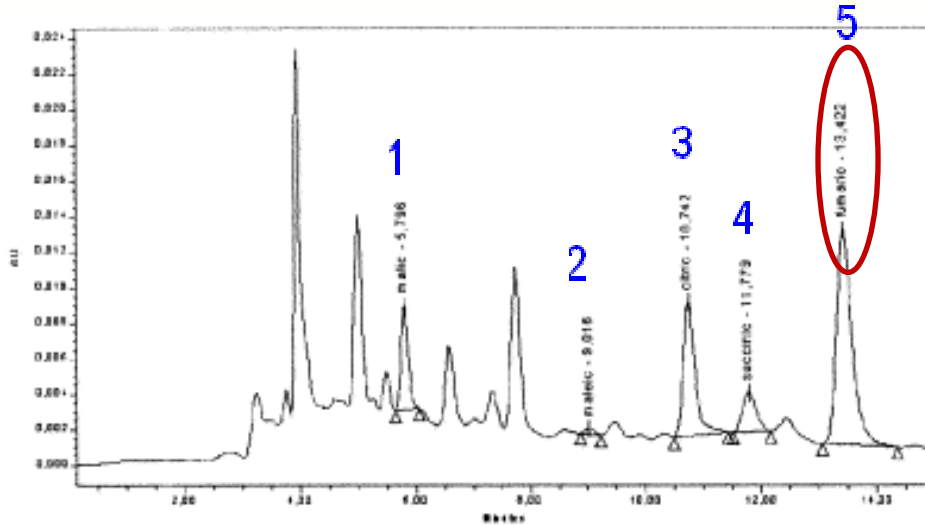
Organic acid	Recovery (%) (mean \pm SD)	RSD (%)
Malic	101,8 \pm 0,18	0,18
Maleic	103,3 \pm 0,099	0,10
Citric	100,8 \pm 0,085	0,08
Succinic	99,2 \pm 0,34	0,34
Fumaric	103,4 \pm 1,43	1,38

Recovery of carboxylic acids added to honey after solid-phase extraction procedure

Organic acid	Recovery (%) (mean \pm SD)	RSD (%)
Malic	62,9 \pm 4,4	7,0
Maleic	93,4 \pm 8,2	8,8
Citric	99,4 \pm 1,5	1,5
Succinic	75,0 \pm 5,0	6,7
Fumaric	94,4 \pm 4,6	4,9

HPLC PROCEDURE

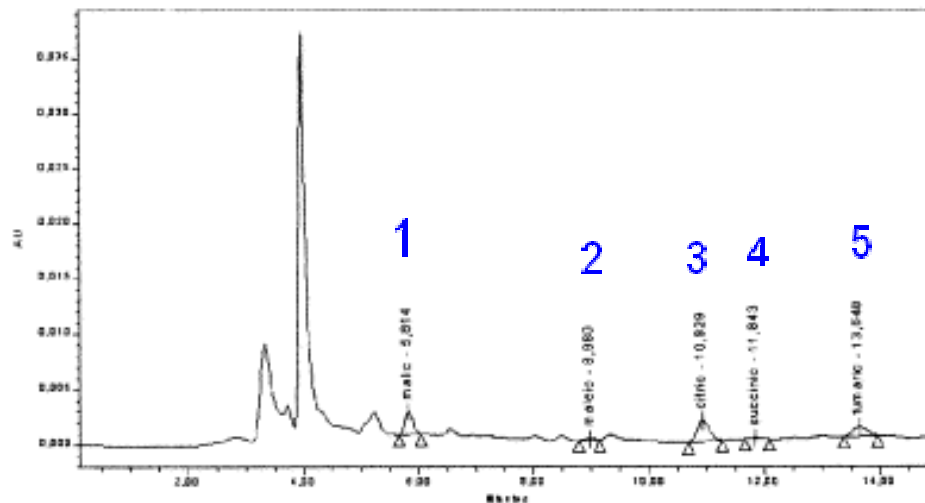
(A)



(A) *Castanea sativa* honey

- 1 Malic acid
- 2 Maleic acid
- 3 Citric acid
- 4 Succinic acid
- 5 Fumaric acid

(B)



(B) Multifloral honey

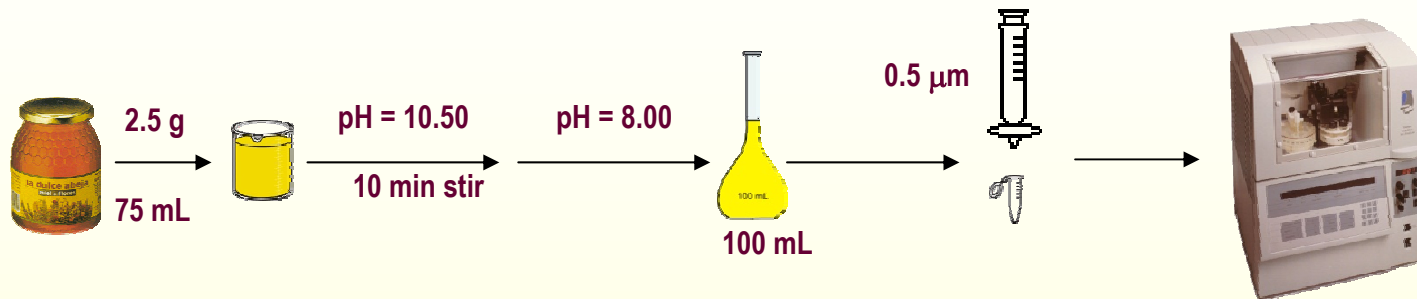


CAPILLARY ZONE ELECTROPHORESIS (CZE)

(Mato et al. 2006. J. Agric. Food Chem. 54)

CAPILLARY ZONE ELECTROPHORESIS

TREATMENT OF THE HONEY SAMPLE



ELECTROPHORETIC CONDITIONS

**Sample injection
(in TRIPLICATE)**

**Hydrodynamic mode (10 cm elevation)
Time of injection: 30 seconds
Quantity of sample: 37.2 nL**

**Electrolyte
composition**

**7.5 mM NaH₂PO₄; 2.5 mM Na₂HPO₄; 2.5 mM TTAOH;
0.24 mM CaCl₂ (pH = 6.40)**

Separation

**Capillary column: 60 cm x 75 µm ID
Temperature: 25 °C
Running voltage: -25 kV**

Detection

UV DIRECT (185 nm)

Time of analysis

4 minutes

CAPILLARY ZONE ELECTROPHORESIS

ORGANIC ACIDS DETERMINED

Oxalic

Formic

Malic

Succinic

Pyruvic

Acetic

Lactic

Citric

Gluconic

CAPILLARY ELECTROPHORESIS

	LOD (mg/kg)	LOQ (mg/kg)	CALIBRATION PLOTS [y=ax+b (r)]	
OXALIC ACID	0.4	12	y = 26.7x + 502	(0.9999)
FORMIC ACID	2.1	23	y = 10.7x + 212	(0.9996)
MALIC ACID	2.0	21	y = 10.8x + 201	(0.9997)
SUCCINIC ACID	2.0	12	y = 13.4x + 294	(0.9998)
PYRUVIC ACID	7.0	39	y = 9.8x + 335	(0.9996)
ACETIC ACID	11	34	y = 17.9x + 233	(0.9999)
LACTIC ACID	4.2	26	y = 9.6x + 204	(0.9999)
CITRIC ACID	9.2	28	y = 12.8x - 533	(0.9996)
GLUCONIC ACID	38	78	y = 5.3x + 56	(0.9999)

LOD= detection limit

LOQ= quantification limit

y= peak area

x= amount of organic acid (mg/kg)

Calibration test: In triplicate

CAPILLARY ZONE ELECTROPHORESIS

Precision

Repeatability

Injection of the honey sample 5 times

Relative standard deviations (RSDs):

0.2% (lactic acid) - 4.6% (formic acid)

Reproducibility

Analysis of each honey sample on 3 different days over 1 month.

Relative standard deviations (RSDs):

0.5% (acetic acid) – 10.0% (oxalic acid)

Recovery

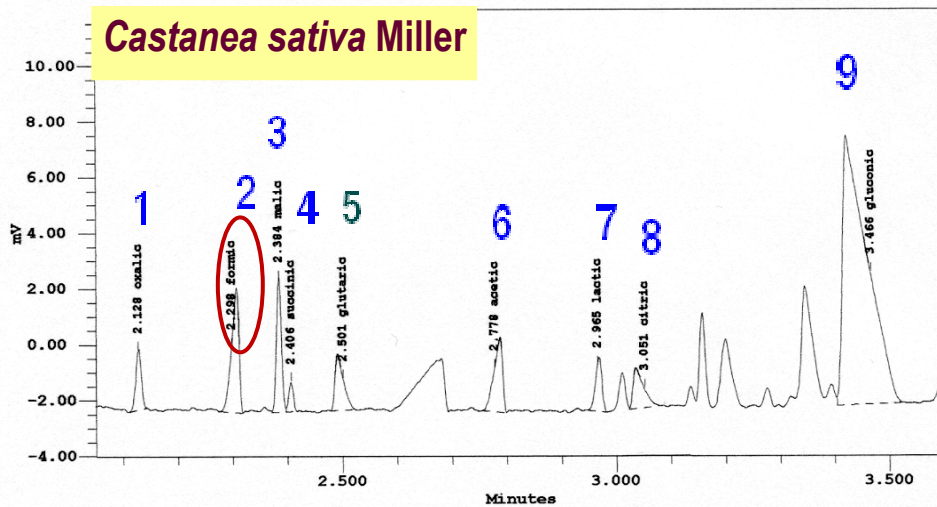
2 honey samples: Low organic acid contents.

Adding three increasing amounts of an organic acid standard mixture to a half amount of honey (1.25 g)

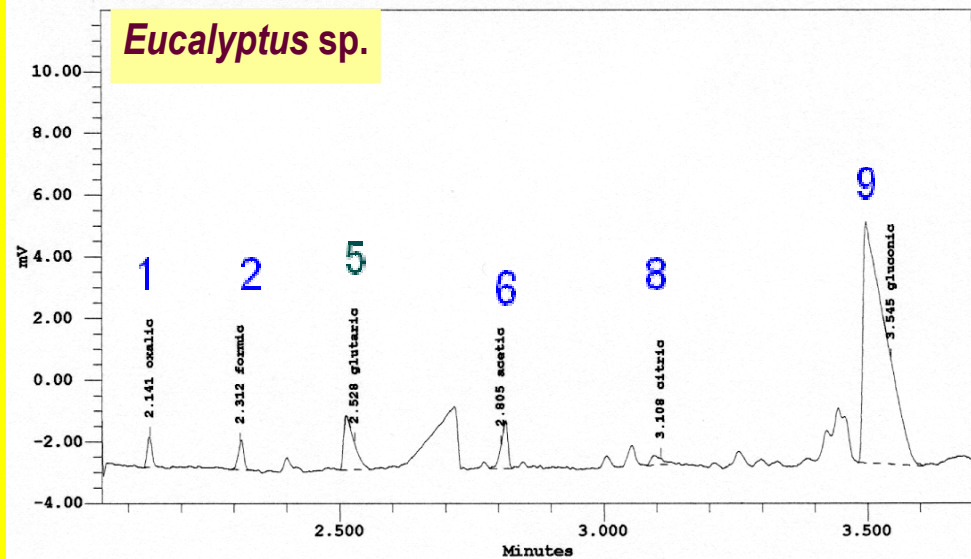
Recoveries (mean (%) \pm SD)

89.4 \pm 10.1 (citric acid) – 104.6 \pm 4.8 (acetic acid)

CAPILLARY ZONE ELECTROPHORESIS



- 1 Oxalic acid
- 2 Formic acid
- 3 Malic acid
- 4 Succinic acid
- 5 Glutaric acid (reference acid to calculate the relative migration times)
- 6 Acetic acid
- 7 Lactic acid
- 8 Citric acid
- 9 Gluconic acid





COMPARISON OF METHODS

NON-AROMATIC ORGANIC ACIDS

Gluconic acid
Enzymatic ← → Capillary electrophoresis

Citric acid
Enzymatic ← HPLC ↓ → Capillary electrophoresis

Malic acid
Enzymatic ← HPLC ↓ → Capillary electrophoresis

Succinic acid
HPLC ← → Capillary electrophoresis

STATISTICAL ANALYSIS

SPSS for Windows v. 10.0.6 SPSS Inc. (1999)

Purposes:

1.- To estimate if the methods (enzymatic, HPLC and CZE) lead to the same results.

Correlation among the results obtained by the three methods.

Comparison of values: *t-test* at a confidence level of 99%.

2.- To propose a method of choice.

GLUCONIC ACID (g/kg)

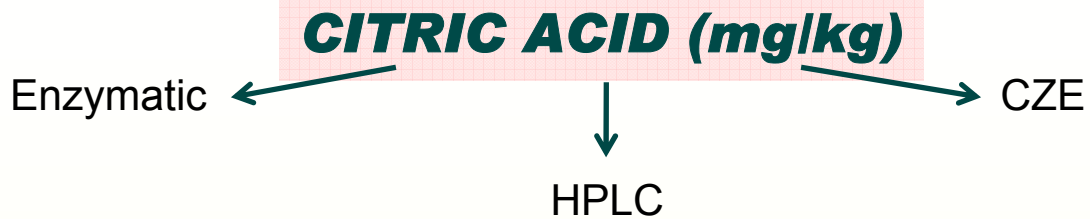
Enzymatic

CZE

- 1) There is a **significant correlation** between the values of gluconic acid between both methods ($r = 0.998$).
- 2) Student's t test showed that **there are not significant differences** ($p > 0.01$) between the results obtained by both the enzymatic method and the CZE procedure.

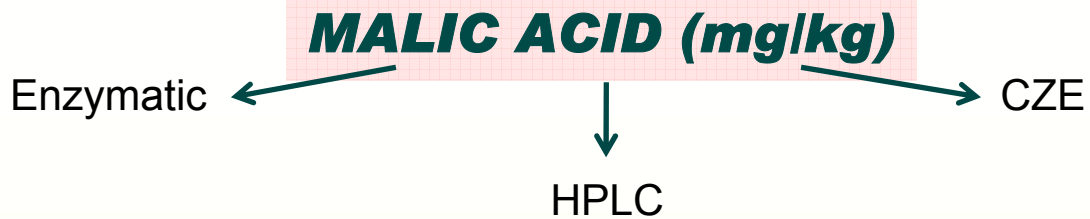
Enzymatic method: HIGHER { specificity
precision and accuracy
sensitivity

CZE method: Numerous organic acids are determined simultaneously



- 1) There is a **significant correlation** among the values of citric acid obtained by the three methods (the lowest r was 0.990). The **best correlation** was obtained between enzymatic and HPLC methods ($r= 0.993$).
- 2) Student's t test showed that **there are not significant differences** ($p > 0.01$) between:
 - All the results obtained by both enzymatic and HPLC methods.
 - All the results obtained by HPLC and CZE methods.
 - The results obtained by both enzymatic and CZE methods, but only for concentrations lower than 250 mg/kg.

Enzymatic method:	HIGHER	}	specificity simplicity precision
HPLC method:	HIGHER	}	sensitivity Other organic acids are determined simultaneously.



- 1) There is a **significant correlation** among the values of malic acid obtained by the three methods (the lowest r was 0.947). The **best correlation** was obtained between enzymatic and CZE methods ($r= 0.999$).
- 2) Student's t test showed that **there are not significant differences** ($p > 0.01$) between the results obtained by the three methods.

Enzymatic method: HIGHER { specificity
sensitivity

CZE method: Numerous organic acids are determined simultaneously

Better { precision
accuracy

SUCCINIC ACID (mg/kg)

HPLC

CZE

There is **no significant correlation** between the values of succinic acid between both methods.

Values of succinic acid are considerably higher by HPLC

Interference

CZE method: { No interferences
Excellent recoveries

CONCLUSIONS

Gluconic acid has been determined by enzymatic and CZE methods. Both methods lead to the same results at a confidence level of 99% for the range of values studied. Enzymatic method provides greater specificity and sensitivity, whereas CZE has the advantage of determining a profile of non-aromatic organic acids. Precision and accuracy have been similar with both methods, but slightly better with the enzymatic procedure.

Citric acid has been determined by enzymatic, HPLC, and CZE methods. All methods are comparable, at a confidence level of 99%, for concentrations lower than 250 mg/kg. For any concentration, the methods of choice would be enzymatic and HPLC. Enzymatic method provides greater specificity, precision and simplicity. HPLC procedure gives higher sensitivity and a profile of other minority non-aromatic organic acids.

Malic acid has been determined by enzymatic, HPLC, and CZE methods. All methods lead to the same results at a confidence level of 99% for the range of values studied. Enzymatic method provides greater specificity and sensitivity, whereas CZE has the advantage of determining a profile of non-aromatic organic acids. Precision and accuracy have been similar with both methods, but slightly better with the CZE procedure.

Succinic acid has been determined by HPLC, and CZE methods, which are not comparable. The method of choice would be CZE, because there are not interferences and the procedure provides excellent recoveries.

MANY THANKS FOR YOUR ATTENTION



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